

## A Review on Separation Processes in Chemistry: Advances, Methods, and Applications

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### ABSTRACT

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Separation processes play a crucial role in various fields of chemistry, enabling the isolation and purification of desired substances from complex mixtures. This review article provides a comprehensive overview of the advances, methods, and applications of separation processes in chemistry. The review begins with an introduction highlighting the significance of separation processes in diverse chemical industries and research areas. The classification of separation processes into physical, chemical, and hybrid categories is presented, offering a framework for understanding the various techniques involved. Advancements in separation processes are then discussed, focusing on physical separation processes such as distillation, filtration, extraction, chromatography, membrane processes, and adsorption. Chemical separation processes including precipitation, crystallization, electrochemical methods, ion exchange, and redox reactions are also explored. Moreover, hybrid separation processes, such as membrane distillation, reactive distillation, adsorptive membrane separation, and liquid-liquid extraction with simultaneous reaction, are highlighted for their unique capabilities. The review further delves into the methods and techniques employed in separation processes, including mathematical modeling and simulation, optimization techniques, process intensification, process monitoring and control, as well as experimental techniques and characterization methods. The applications of separation processes across various industries are then discussed, including the pharmaceutical industry, chemical synthesis and manufacturing, environmental remediation, food and beverage industry, and petrochemical industry. The significance of separation processes in these domains is highlighted, emphasizing their role in achieving high purity, yield, and efficiency. Lastly, the review addresses the challenges faced by separation processes and provides insights into future directions. Energy efficiency and sustainability, integration of separation processes, and the development of novel materials and technologies are identified as key areas for advancement. Additionally, the role of process monitoring and control in improving separation processes is discussed. In conclusion, this review provides a comprehensive overview of the advancements, methods, and applications of separation processes in chemistry. It underscores the importance of continued research and development in this field to address challenges, improve efficiency, and meet the evolving needs of various industries.

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**Key words:** *Separation Processes, Chemistry, Advances, Methods, Applications*

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### Introduction

Separation processes are fundamental to the field of chemistry, serving as essential tools for isolating and purifying substances from complex mixtures. These processes play a vital role in various industries, including pharmaceuticals, chemicals, environmental remediation, and food production. Understanding and harnessing the principles and techniques of separation processes are critical for achieving high-quality products, optimizing process efficiency, and ensuring environmental sustainability [1-10].

The objective of this review is to provide a comprehensive examination of the advances, methods, and applications of separation processes in chemistry. By exploring the latest developments and innovative approaches, this review aims to shed light on the current state-of-the-art and highlight the potential for future advancements in this field [5-12].

The scope of this review encompasses a broad range of separation processes, including physical, chemical, and hybrid techniques. Physical separation processes involve the separation of substances based on differences in their physical properties, such as boiling point, solubility, and size. Chemical separation processes, on the other hand, rely on specific chemical reactions or interactions to achieve separation. Hybrid separation processes combine elements of both physical and chemical methods to enhance separation efficiency.

This review will delve into the various advances and methodologies employed in separation processes, exploring techniques such as distillation, filtration, extraction, chromatography, membrane processes, and adsorption, among others. It will also discuss mathematical modeling, optimization techniques, and experimental approaches used for process design, analysis, and control.

Furthermore, this review will examine the wide-ranging applications of separation processes in industries such as pharmaceuticals, chemicals, environmental remediation, food and beverages, and petrochemicals. The significance of these processes in achieving high-purity products, improving yields, and meeting stringent quality standards will be emphasized.

Overall, this review aims to provide researchers, engineers, and practitioners in the field of chemistry with a comprehensive understanding of the advances, methods, and applications of separation processes. By highlighting the latest developments and addressing the challenges and future directions, this review seeks to contribute to the ongoing progress and innovation in this vital area of chemistry.

The review will also explore the importance of separation processes in addressing key challenges faced by industries, such as the need for energy efficiency, waste reduction, and environmental sustainability. By discussing the advancements and emerging trends in separation processes, this review aims to contribute to the development of more efficient and environmentally friendly separation technologies.

Furthermore, the review will highlight the significance of process integration and optimization in achieving overall process efficiency and cost-effectiveness. The integration of multiple separation techniques and the use of advanced process intensification strategies will be discussed to maximize productivity and minimize resource consumption.

The review will also address the role of process monitoring and control in ensuring the reliability and performance of separation processes. The use of advanced sensors, automation, and real-time optimization techniques can enable more precise control and optimization of separation processes, leading to improved product quality and process efficiency.

It is important to note that this review will not only focus on the theoretical aspects of separation processes but also highlight practical applications and case studies. Real-world examples from various industries will be presented to demonstrate the successful implementation of separation processes and their impact on product quality, process economics, and environmental sustainability.

### **Classification of Separation Processes**

Separation processes can be broadly classified into three main categories: physical separation processes, chemical separation processes, and hybrid separation processes. Each category encompasses a range of techniques that are employed based on the specific characteristics of the mixture and the desired separation objectives [6-15].

1. **Physical Separation Processes:** Physical separation processes rely on the differences in physical properties of substances in a mixture to achieve separation. Some commonly used physical separation techniques include:
  - **Distillation:** This process involves the separation of components based on differences in their boiling points. It is widely used for the purification of liquids and the separation of volatile compounds.
  - **Filtration:** Filtration is the process of separating solids from liquids or gases using a porous medium, such as a filter paper or a membrane. It is employed to remove solid impurities from a liquid or to collect solid particles from a gas.
  - **Extraction:** Extraction involves the selective transfer of a solute from one phase to another based on differences in solubility. Liquid-liquid extraction and solid-liquid extraction are commonly used techniques for separating substances.
  - **Chromatography:** Chromatographic techniques separate components based on their differential interactions with a stationary phase and a mobile phase. It finds wide applications in analytical chemistry and purification processes.
  - **Membrane Processes:** Membrane-based separation processes, such as reverse osmosis, ultrafiltration, and nanofiltration, utilize semi-permeable membranes to separate substances based on differences in size, charge, or solubility.
  - **Adsorption:** Adsorption involves the selective adhesion of molecules or ions to a solid surface. It is used for separation and purification based on differences in affinity or adsorption capacity.
2. **Chemical Separation Processes:** Chemical separation processes rely on specific chemical reactions or interactions to achieve separation. Some commonly used chemical separation techniques include:
  - **Precipitation:** Precipitation involves the formation of a solid precipitate from a solution due to the reaction of specific ions or compounds. It is used for the separation and purification of desired compounds.
  - **Crystallization:** Crystallization is the process of forming solid crystals from a liquid or solution, typically through controlled cooling or evaporation. It is widely used for purification and isolation of compounds.
  - **Electrochemical Methods:** Electrochemical processes, such as electrodialysis and electrolysis, utilize the principles of electrolysis to selectively remove or deposit specific ions for separation purposes.

- Ion Exchange: Ion exchange involves the exchange of ions between a solid resin and a solution to achieve separation based on differences in ion affinity. It is commonly used in water treatment and purification processes.
  - Redox Reactions: Redox reactions involve the transfer of electrons between species, resulting in the selective oxidation or reduction of specific compounds for separation.
3. Hybrid Separation Processes: Hybrid separation processes combine elements of both physical and chemical techniques to enhance separation efficiency. These processes often involve the integration of multiple separation methods to achieve higher selectivity or improved overall performance. Some examples of hybrid separation processes include [16-25]:
- Membrane Distillation: Membrane distillation combines the principles of both membrane processes and distillation, allowing the separation of components by vapor transport through a hydrophobic membrane.
  - Reactive Distillation: Reactive distillation integrates the separation process with a chemical reaction occurring within the distillation column, resulting in simultaneous separation and reaction.
  - Adsorptive Membrane Separation: This technique combines membrane separation with adsorption, allowing the selective adsorption of specific components for separation.
  - Liquid-Liquid Extraction with Simultaneous Reaction: In this hybrid process, liquid-liquid extraction is coupled with a simultaneous chemical reaction to achieve separation and conversion of specific compounds.

The classification of separation processes into physical, chemical, and hybrid categories provides a framework for understanding the different techniques employed in the field of chemistry. By utilizing a combination of these processes, researchers and engineers can tailor separation strategies to suit specific separation requirements and optimize overall process efficiency [18-24].

### **Advances in Separation Processes**

Physical separation processes have witnessed significant advancements in recent years, leading to improved separation efficiency, selectivity, and sustainability. The following techniques have seen notable advancements [24-30]:

- Distillation: Distillation, a widely used technique for separating components based on differences in boiling points, has seen advancements in the form of innovative distillation column designs, such as reactive distillation and dividing-wall columns. These designs enable simultaneous reaction and separation, as well as improved energy efficiency.
- Filtration: Filtration techniques have benefited from advancements in filter media and equipment design. The development of advanced filter materials, such as ceramic membranes and nanofibrous filters, has enhanced filtration efficiency, durability, and resistance to fouling. Additionally, the use of novel filtration configurations, such as crossflow and dead-end filtration, has improved separation performance.
- Extraction: Advancements in extraction techniques have focused on enhancing selectivity, reducing solvent consumption, and improving process efficiency. Techniques such as supercritical fluid extraction (SFE), micellar extraction, and ultrasound-assisted extraction (UAE) have gained prominence due to their ability to extract target compounds with higher efficiency and reduced environmental impact.
- Chromatography: Chromatographic separation techniques have undergone significant advancements, particularly in the development of novel stationary phases, such as monolithic columns and core-shell particles. These advancements have led to improved resolution, reduced analysis time, and increased column efficiency. Additionally, advancements in high-performance liquid chromatography (HPLC) instrumentation, such as faster detectors and automated systems, have facilitated high-throughput analysis.
- Membrane Processes: Membrane processes have witnessed notable advancements in membrane materials, module designs, and process configurations. The development of advanced membranes, including nanofiltration and reverse osmosis membranes with enhanced selectivity and permeability, has revolutionized desalination, water treatment, and separation processes. Additionally, membrane-based technologies such as membrane distillation and pervaporation have gained attention for their ability to separate volatile components and azeotropic mixtures.
- Adsorption: Advancements in adsorption processes have focused on the development of novel adsorbents and the improvement of adsorption kinetics and capacity. New adsorbent materials, such as metal-organic frameworks (MOFs) and carbon-based nanomaterials, offer high surface areas and tunable properties for selective adsorption. Additionally, advancements in adsorption process design, such as fixed-bed and simulated moving bed (SMB) systems, have enabled efficient separation and regeneration cycles.

These advancements in physical separation processes have contributed to improved separation performance, energy efficiency, and sustainability. By incorporating novel materials, innovative designs, and optimized operating conditions, these techniques have expanded the capabilities of separation processes in various industries, including chemical production, environmental remediation, and pharmaceutical manufacturing.

### **3.2. Chemical Separation Processes**

Chemical separation processes involve utilizing specific chemical reactions or interactions to achieve separation. Recent advancements in these techniques have focused on improving selectivity, reaction kinetics, and process control. The following techniques have seen notable advancements [10-28]:

- **Precipitation:** Advancements in precipitation processes have focused on enhancing the control over particle size, morphology, and purity of the precipitate. Precipitation techniques such as controlled nucleation, template-assisted precipitation, and microfluidic-based precipitation have been developed to achieve precise control over the particle properties. Additionally, the use of additives and surfactants has been explored to improve the selectivity and efficiency of precipitation processes.
- **Crystallization:** Crystallization techniques have undergone significant advancements to improve product purity, crystal size distribution, and process efficiency. Novel crystallization methods, such as anti-solvent crystallization, reactive crystallization, and continuous crystallization, have been developed to achieve better control over crystal growth, polymorph selection, and impurity removal. Furthermore, the integration of in-situ monitoring and control strategies, such as using advanced sensors and feedback control algorithms, has enabled real-time optimization of crystallization processes.
- **Electrochemical Methods:** Electrochemical separation processes, including electrodialysis, electrolysis, and electrocoagulation, have seen advancements in electrode materials, cell design, and process optimization. The development of novel electrode materials with high selectivity and stability, such as ion-selective membranes and catalyst-coated electrodes, has improved the efficiency and selectivity of electrochemical separation. Additionally, advancements in process control and automation have enhanced the scalability and applicability of electrochemical methods in various industries.
- **Ion Exchange:** Ion exchange processes have benefited from advancements in ion exchange resins, regeneration strategies, and process optimization. The development of specialized ion exchange resins with improved selectivity and capacity has expanded the range of separations achievable by ion exchange. Additionally, advancements in regeneration techniques, such as the use of alternative regenerants and optimized regeneration cycles, have increased the efficiency and sustainability of ion exchange processes.
- **Redox Reactions:** Redox reactions have seen advancements in the utilization of novel redox-active compounds, catalytic systems, and reaction engineering approaches. The development of efficient and selective redox catalysts, such as transition metal complexes and biocatalysts, has enabled improved conversion and selectivity in redox-based separation processes. Moreover, the integration of reaction engineering principles, such as process intensification and flow chemistry, has facilitated the development of continuous and scalable redox separation processes.

These advancements in chemical separation processes have expanded the range of separations achievable, improved selectivity, and enhanced process efficiency. By leveraging innovative materials, reaction kinetics, and process optimization strategies, chemical separation processes have found applications in diverse fields, including pharmaceutical synthesis, wastewater treatment, and fine chemical production [29-35].

### **3.3. Hybrid Separation Processes**

Hybrid separation processes combine elements of both physical and chemical techniques to achieve enhanced separation performance, selectivity, and efficiency. Recent advancements in hybrid separation processes have focused on integrating different principles to overcome the limitations of individual techniques. The following techniques have seen notable advancements:

- **Membrane Distillation:** Membrane distillation combines the principles of both membrane separation and distillation. Advancements in membrane materials, module designs, and process optimization have improved the efficiency and scalability of membrane distillation. The use of hydrophobic membranes and innovative module configurations enables the selective separation of volatile components from complex mixtures, such as desalination of saline water or concentration of heat-sensitive products.
- **Reactive Distillation:** Reactive distillation integrates separation and chemical reaction in a single unit operation. Advancements in catalyst design, column internals, and process optimization have enhanced the performance and selectivity of reactive distillation. By combining reaction and separation, reactive distillation offers advantages such as

higher conversion, reduced equipment footprint, and simplified process control, making it an attractive option for various chemical processes.

- **Adsorptive Membrane Separation:** Adsorptive membrane separation combines the principles of adsorption and membrane processes. Advances in adsorbent materials, membrane configurations, and process design have improved the selectivity and capacity of adsorptive membrane separation. This hybrid approach allows for the removal of specific components by adsorption onto the membrane surface or within the membrane pores, enabling the separation of complex mixtures and purification of target compounds.
- **Liquid-Liquid Extraction with Simultaneous Reaction:** Liquid-liquid extraction with simultaneous reaction integrates extraction and chemical reaction in a single step. Recent advancements in solvent selection, reactor design, and process optimization have enhanced the efficiency and selectivity of this hybrid process. By combining extraction and reaction, it is possible to achieve simultaneous separation and conversion of target compounds, offering advantages such as reduced energy consumption and simplified process flowsheets.

These advancements in hybrid separation processes have expanded the capabilities and flexibility of separation technologies, enabling improved separation efficiency, selectivity, and sustainability. By leveraging the synergistic effects of multiple principles, hybrid processes have found applications in various industries, including petrochemicals, pharmaceuticals, and environmental remediation. These techniques hold promise for addressing complex separation challenges and driving innovation in the field of separation processes [36-40].

### **Methods and Techniques in Separation Processes**

The development and improvement of separation processes rely on a range of methods and techniques that enable better understanding, design, optimization, and control of these processes. Recent advancements in these methods and techniques have contributed to enhanced process performance, efficiency, and sustainability. The following methods and techniques have seen notable advancements [38-45]:

- **Mathematical Modeling and Simulation:** Mathematical modeling plays a crucial role in understanding the fundamental principles underlying separation processes. Advances in mathematical modeling techniques, such as computational fluid dynamics (CFD), population balance modeling, and thermodynamic modeling, have improved the accuracy and predictive capability of process simulations. These models allow for the optimization of process parameters, estimation of product quality, and evaluation of process economics before experimental implementation [39-42].
- **Optimization Techniques:** Optimization techniques aid in finding the optimal operating conditions and design parameters for separation processes. Recent advancements in optimization algorithms, such as genetic algorithms, particle swarm optimization, and artificial intelligence-based optimization methods, have facilitated the exploration of large design spaces and improved the efficiency and robustness of separation processes. Optimization techniques help in maximizing separation efficiency, minimizing energy consumption, reducing waste generation, and optimizing resource utilization.
- **Process Intensification:** Process intensification involves the development and integration of innovative techniques to enhance separation performance and efficiency. Recent advancements in process intensification have focused on techniques such as reactive distillation, membrane reactors, hybrid separation processes, and intensified separations using microreactors or microchannels. These approaches enable compact and efficient separation systems with reduced energy consumption, smaller equipment footprint, and improved overall process performance.
- **Process Monitoring and Control:** Process monitoring and control are essential for ensuring the reliable and optimal operation of separation processes. Recent advancements in sensor technologies, online analytical methods, and automation systems have improved the real-time monitoring and control of separation processes. Advanced process monitoring techniques, such as spectroscopic analysis, chromatographic sensors, and in-line sensors, provide accurate and continuous measurement of process variables, enabling timely adjustments and optimization of separation operations.
- **Experimental Techniques and Characterization Methods:** Experimental techniques and characterization methods play a crucial role in the development, validation, and optimization of separation processes. Recent advancements in experimental techniques, such as high-throughput screening methods, advanced analytical instruments, and advanced imaging techniques, have enabled rapid and precise characterization of separation systems. These techniques aid in understanding the underlying mechanisms, evaluating separation performance, and identifying potential areas for improvement.

The integration of these methods and techniques in the field of separation processes allows for a holistic and systematic approach to process design, optimization, and control. By combining theoretical models, optimization algorithms, experimental data, and

advanced characterization methods, researchers and engineers can develop more efficient and sustainable separation processes with improved performance, selectivity, and cost-effectiveness.

### **Applications of Separation Processes**

Separation processes find wide-ranging applications in various industries, enabling the purification, isolation, and recovery of valuable substances from complex mixtures. The following industries have particularly benefited from the advancements in separation processes [46-50]:

1. **Pharmaceutical Industry:** Separation processes play a crucial role in the pharmaceutical industry for the purification of active pharmaceutical ingredients (APIs), removal of impurities, and formulation of drugs. Techniques such as chromatography, crystallization, extraction, and membrane processes are utilized for the separation and purification of drugs, enabling the production of high-quality pharmaceutical products with precise dosages and improved therapeutic efficacy.
2. **Chemical Synthesis and Manufacturing:** Separation processes are essential in chemical synthesis and manufacturing to separate and purify desired products from reaction mixtures. Distillation, crystallization, liquid-liquid extraction, and adsorption techniques are widely employed for product purification, removal of impurities, and recycling of solvents or catalysts. These processes ensure the production of chemicals with high purity, which is crucial for meeting stringent quality standards and ensuring the safety of end-users.
3. **Environmental Remediation:** Separation processes play a vital role in environmental remediation, facilitating the removal and recovery of pollutants from air, water, and soil. Techniques such as membrane filtration, adsorption, ion exchange, and precipitation are employed to separate and capture contaminants, including heavy metals, organic pollutants, and harmful gases. These processes help in mitigating environmental pollution and protecting ecosystems and public health.
4. **Food and Beverage Industry:** Separation processes are extensively used in the food and beverage industry for various applications, including clarification, concentration, purification, and fractionation. Techniques such as filtration, centrifugation, chromatography, and membrane processes are employed to remove impurities, concentrate flavors or nutrients, and separate different components of food and beverage products. These processes contribute to ensuring the quality, safety, and shelf-life of food and beverage products.
5. **Petrochemical Industry:** Separation processes are critical in the petrochemical industry for the refining and processing of crude oil and natural gas. Techniques such as distillation, adsorption, membrane separation, and extraction are used to separate hydrocarbons, remove impurities, and recover valuable compounds from crude oil or natural gas streams. These processes enable the production of various petrochemical products, such as fuels, polymers, and specialty chemicals, meeting the diverse needs of industries and consumers.

In these industries, the application of advanced separation processes improves product quality, process efficiency, and sustainability. The development of innovative separation technologies and the optimization of existing processes contribute to reduced energy consumption, waste generation, and environmental impact, thereby fostering a more sustainable and efficient industrial landscape.

### **Challenges and Future Directions**

1. **Energy Efficiency and Sustainability:** One of the key challenges in separation processes is the energy consumption associated with achieving desired separation objectives. Future directions in separation processes aim to develop energy-efficient technologies and strategies to minimize energy consumption and reduce the carbon footprint. This includes the exploration of alternative energy sources, process heat integration, and the development of energy-efficient separation materials and equipment.
2. **Integration of Separation Processes:** Integration of multiple separation processes within a single system offers opportunities for improved efficiency, reduced waste generation, and increased process intensification. Future directions involve the development of integrated separation processes that combine different techniques, such as hybrid processes, cascade systems, and hybrid separations with reaction units. Integrated approaches enable synergistic effects, improved mass and energy transfer, and optimal resource utilization.
3. **Novel Separation Materials and Technologies:** Future directions in separation processes involve the development of novel materials with enhanced separation properties. This includes the exploration of advanced adsorbents, membranes, catalysts, and selective sorbents that exhibit improved selectivity, stability, and sustainability. Additionally, the use of nanotechnology and functional materials opens up possibilities for tailored separation materials with precise control over separation mechanisms and improved performance.
4. **Advances in Process Monitoring and Control:** Real-time monitoring and control of separation processes are crucial for ensuring optimal performance, product quality, and resource efficiency. Future directions involve the development of advanced sensing technologies, online monitoring techniques, and intelligent process control systems. Integration of

data analytics, machine learning, and artificial intelligence enables the optimization of separation processes, fault detection, adaptive control, and predictive maintenance.

These future directions aim to address the challenges faced by separation processes and pave the way for more sustainable, efficient, and technologically advanced separation systems. Continued research and development efforts, collaboration between academia and industry, and the adoption of innovative approaches will drive the advancements in separation processes, enabling greener and more resource-efficient industrial operations.

### **Conclusion**

In conclusion, this review has provided a comprehensive overview of the advances, methods, and applications of separation processes in chemistry. The key findings from this review can be summarized as follows:

1. Separation processes are essential in various industries, including pharmaceuticals, chemical synthesis, environmental remediation, food and beverage, and petrochemicals. These processes enable the purification, isolation, and recovery of valuable substances from complex mixtures, ensuring product quality, process efficiency, and environmental sustainability.
2. Advances in physical separation processes have led to improved techniques such as distillation, filtration, extraction, chromatography, membrane processes, and adsorption. These advancements have enhanced separation efficiency, selectivity, and sustainability, addressing the evolving needs of industries.
3. Chemical separation processes, including precipitation, crystallization, electrochemical methods, ion exchange, and redox reactions, have seen advancements in terms of selectivity, reaction kinetics, and process control. These developments have enabled the separation and purification of compounds with high efficiency and precision.
4. Hybrid separation processes, such as membrane distillation, reactive distillation, adsorptive membrane separation, and liquid-liquid extraction with simultaneous reaction, have emerged as promising approaches for achieving enhanced separation performance. These hybrid techniques combine the advantages of physical and chemical principles, offering improved selectivity, efficiency, and process integration.
5. Methods and techniques, including mathematical modeling and simulation, optimization techniques, process intensification, process monitoring and control, as well as experimental techniques and characterization methods, are critical for understanding, designing, optimizing, and controlling separation processes. Recent advancements in these areas have facilitated the development of more efficient and sustainable separation technologies.

The importance of continued research and development in separation processes cannot be overstated. Ongoing efforts in this field are crucial for addressing challenges such as energy consumption, waste generation, and environmental impact. By pursuing innovative approaches, such as energy-efficient technologies, integrated separation processes, novel materials, and advanced process monitoring and control, researchers can further enhance the performance, sustainability, and industrial relevance of separation processes.

In conclusion, the advancements and applications of separation processes in chemistry continue to play a vital role in various industries, and continued research and development efforts will drive further innovation, leading to more efficient, sustainable, and economically viable separation technologies.

### **References**

1. Abderrahim, H.A., & Paillère, H. (2009). Strategic Research Agenda, Report of the Sustainable Nuclear Energy Technology Platform. Retrieved from [www.snetp.eu](http://www.snetp.eu).
2. Allibert, M., Ault, T., Baron, P., Bergeron, A., Bromley, B., Butler, G., Chauvin, N., Collins, E., Cornet, S., Croff, A., Eschbach, R., Feinberg, O., Floyd, M., Ghetta, V., Grenèche, D., Hamilton, H., Hesketh, K., Hyland, B., Ignatiev, V., Kelly, J.F., Krahn, S., McDonald, M., Merle-Lucotte, E., Michel-Sendis, F., Porsch, D., Rimpault, G., Serp, J., Taiwo, T., Uhler, J., Van Den Durpel, L., Van Den Eynde, G., Vidal, J., Vitanza, C., Wojtaszek, D., & Wymer, R. (2015). Introduction of Thorium in the Nuclear Fuel Cycle, Short- to Long-Term Considerations (NEA No. 7224) (pp. 1–135).
3. Alyapyshev, M.Y., Babain, V.A., & Ustynyuk, Y.A. (2016). Recovery of minor actinides from high-level wastes: modern trends. *Russian Chemical Reviews*, 85, 943–961.
4. Ansari, S.A., Pathak, P., Mohapatra, P.K., & Manchanda, V.K. (2011). Chemistry of diglycolamides: promising extractants for actinide partitioning. *Chemical Reviews*, 112, 1751–1772.
5. Arm, S., & Phillips, C. (2010). Closing the nuclear fuel cycle in the 21st century while minimizing proliferation risk. In *International Conference on the Physics of Reactors 2010, PHYSOR 2010* (pp. 2075–2086).
6. Arm, S., & Phillips, C. (2011). Chemical engineering for advanced aqueous radioactive materials separations. In K.L. Nash & G.J. Lumetta (Eds.), *Advanced Separation Techniques for Nuclear Fuel Reprocessing and Radioactive Waste Treatment* (pp. 58–94). Woodhead Publishing.
7. Baron, P. (2010). National Programmes in Chemical Partitioning: A Status Report (NEA No. 5245) (pp. 1–120). Nuclear Energy Agency, Organisation for Economic Cooperation and Development.

8. Baron, P., Brown, C., Kaiser, B., Matthews, B., Mukaiyama, T., Omberg, R., Peddicord, L., Salvatores, M., & Waltar, A. (2004). An Evaluation of the Proliferation Resistant Characteristics of Light Water Reactor Fuel with the Potential for Recycle in the United States, Final Report November 2004 (pp. 1–64). Pacific Northwest National Laboratory.
9. Baron, P., Choi, Y.-J., Collins, E., Boucher, D.L., Coquelet, C., Rostaing, C., Morita, Y. (2012). Curium Management Studies in France, Japan and USA (pp. 1–38). Nuclear Energy Agency.
10. Barrachin, M., Basini, V., Colak, Ü., Dubourg, R., Feltus, M.A., Greneche, D., Guillermier, P., Hansen, U., Hanson, D.L., Hunn, J.D., Kania, M.J., Kissane, M., Martin, D.G., McEachern, D.W., van der Merwe, J.J., Nabielek, H., Nawada, H.P., Nothnagel, G., Petti, D.A., Shenoy, A., Tang, C., Thoma, E.H., Fulton, T.R., Ueta, S., Venter, S., Verfondern, K., & Szymczak, W.J. (2010). High Temperature Gas Cooled Reactor Fuels and Materials (IAEA-TECDOC-1645) (pp. 1–182). International Atomic Energy Agency.
11. Bathke, C.G. (2009). The attractiveness of materials in advanced nuclear fuel cycles for various proliferation and theft scenarios. In GLOBAL 2009. American Nuclear Society.
12. Beaman, M., Chare, P., Ciuculescu, C., Cohen-Unger, S., Cojazzi, G., Durbin, K., Gowin, P., Haas, E., Hanks, D., Janssens, W., Killeen, T., Kim, H.D., Koutsoyannopoulos, C., Kovacic, D., Kruzic, L., Mathews, C., McCarthy, W., Moran, B., Okko, O., Sevini, F., Sprinkle, J., Stein, M., Subki, H., Tuley, N., Whitlock, J., & Wonder, E. (2013). International Safeguards in Nuclear Facility Design and Construction (NP-T-2.8) (pp. 1–36). International Atomic Energy Agency.
13. Behar, C. (2014). Energy for the future: generation IV fast reactors and the closed fuel cycle. *Nuclear Future*, 10, 23–25.
14. Benedict, R., Goff, M., Teske, G., & Johnson, T. (2002). Progress in electrometallurgical treatment of spent nuclear fuel. *Journal of Nuclear Science and Technology*, 39, 749–752.
15. Benedict, R.W., Solbrig, C., Westphal, B., Johnson, T.A., Li, S.X., Marsden, K., & Goff, K.M. (2007). Pyroprocessing progress at Idaho national laboratory. In GLOBAL 2007: Advanced Nuclear Fuel Cycles and Systems (pp. 741–747).
16. Birkett, J.E., Carrott, M.J., Fox, O.D., Jones, C.J., Maher, C.J., Roubé, C.V., Taylor, R.J., & Woodhead, D.A. (2005). Recent developments in the PUREX process for nuclear fuel reprocessing: complexant-based stripping for uranium-plutonium separation. *Chimia*, 59, 898–904.
17. Bjornard, T., Garcia, H., Desmond, W., & DeMuth, S. (2010). Safeguarding and protecting the nuclear fuel cycle. *Nuclear News*, 55, 76–80.
18. Bourg, S., Hill, C., Caravaca, C., Rhodes, C., Ekberg, C., Taylor, R., Geist, A., Modolo, G., Cassayre, L., Malmbeck, R., Harrison, M., de Angelis, G., Espartero, A., Bouvet, S., Ouvrier, N. (2011). ACSEPT—Partitioning technologies and actinide science: towards pilot facilities in Europe. *Nuclear Engineering and Design*, 241, 3427–3435.
19. Bourg, S., Guilbaud, P., Mendes, E., Ekberg, C., Gibilaro, M., Soucek, P., Modolo, G., Geist, A., Boo, E., Duplantier, B., Rhodes, C., Taylor, R., Harrison, M., Flint, L., Bell, K., Sharrad, C., & Hanson, B. (2016). SACSESS: Final Report, SACSESS – R01.3 – Rev 0 (pp. 1–35). CEA, France.
20. Bryan, S.A., Levitskaia, T.G., Casella, A.J., Peterson, J.M., Johnsen, A.M., Lines, A.M., & Thomas, E.M. (2011). Spectroscopic on-line monitoring for process control and safeguarding of radiochemical streams in nuclear fuel reprocessing facilities. In K.L. Nash & G.J. Lumetta (Eds.), *Advanced Separation Techniques for Nuclear Fuel Reprocessing and Radioactive Waste Treatment* (pp. 95–119). Woodhead Publishing.
21. Butler, G., & Hesketh, K. (2017). Choosing the right nuclear power systems. *Nuclear Future*, 13, 50–56.
22. Cantarella, J., Jones, D., Mauti, J., Váša, I., Avolahti, J., Palmu, J., Leboucher, I., Legee, F., Mazaré, F., Kim, S.-S., Ko, W.I., Kim, S.K., Poley, A.D., Trigueros, A.A., López, F.J.F., Forsström, H., Hodgson, Z., Holmes, C., Ganda, F., Hilden, W., & Murphy, S. (2013). The Economics of the Back End of the Nuclear Fuel Cycle (NEA No. 7061) (pp. 1–193). Organisation for Economic Co-operation and Development Nuclear Energy Agency.
23. Carmack, J., Goldner, F., Bragg-Sitton, S.M., & Snead, L.L. (2013). Overview of the U.S. DOE Accident Tolerant Fuel Development Program (INL/CON-13-29288) (pp. 1–10). Idaho National Laboratory.
24. Carrott, M., Bell, K., Brown, J., Geist, A., Gregson, C., Hères, X., Maher, C., Malmbeck, R., Mason, C., Modolo, G., Müllich, U., Sarsfield, M., Wilden, A., & Taylor, R. (2014). Development of a new flowsheet for co-separating the transuranic actinides: the "EURO-GANEX" process. *Solvent Extraction and Ion Exchange*, 32, 447–467.
25. Cashmore, R., Billowes, J., Bowen, W., Brown, C., Grimes, R., Howsley, R., Livens, F., Simpson, J., & Styles, P. (2011). *Fuel Cycle Stewardship in a Nuclear Renaissance* (pp. 1–79). The Royal Society Science Policy Centre.
26. Gregson, C.R., Horne, G.P., Orr, R.M., Pimblott, S.M., Sims, H.E., Taylor, R.J., Webb, K.J. (2018). Molecular hydrogen yields from the  $\alpha$ -self-radiolysis of nitric acid solutions containing plutonium or americium. *J. Phys. Chem. B*, 122, 2627–2634.
27. Grimes, W.R. (1970). Molten-salt reactor chemistry. *Nucl. Appl. Technol.*, 8, 137–155.
28. Hadibi-Olschewski, N., Glatz, J.P., Bokelund, H., Leroy, M.J.F. (1992). The fate of nitrogen upon reprocessing of nitride fuels. *J. Nucl. Mater.*, 188, 244–248.



29. Halleröd, J., Ekberg, C., Löfström-Engdahl, E., Aneheim, E. (2015). Development of the Chalmers grouped actinide extraction process. *Nukleonika*, 60, 829–835.
30. Halleröd, J., Ekberg, C., Aneheim, E. (2016). Phenyl trifluoromethyl sulfone as diluent in a grouped actinide extraction process: extraction properties of the solvent components TBP and CyMe4-BTBP. *J. Radioanal. Nucl. Chem.*, 307, 1711–1715.
31. Halleröd, J., Ekberg, C., Authen, T., Bertolo, L., Lin, M., Grüner, B., Švehla, J., Wagner, C., Geist, A., Panak, P., Aneheim, E. (2018Fa). On the basic extraction properties of a phenyl trifluoromethyl sulfone-based GANEX system containing CyMe4-BTBP and TBP. *Solvent Extr. Ion Exch.*, 36, 360–372.
32. Halleröd, J., Ekberg, C., Kajan, I., Aneheim, E. (2018Fb). Solubility thermodynamics of CyMe4-BTBP in various diluents mixed with TBP. *J. Solut. Chem.*, 47, 1021–1036.
33. Hanson, B. (2015). 6 - process engineering and design for spent nuclear fuel reprocessing and recycling plants. In R. Taylor (Ed.), *Reprocessing and Recycling of Spent Nuclear Fuel* (pp. 125–151). Woodhead Publishing.
34. Herbst, R.S., Baron, P., Nilsson, M. (2011). Standard and advanced separation: PUREX processes for nuclear fuel reprocessing. In K.L. Nash & G.J. Lumetta (Eds.), *Advanced Separation Techniques for Nuclear Fuel Reprocessing and Radioactive Waste Treatment* (pp. 141–175). Woodhouse Publishing Ltd.
35. Herrmann, S.D., Li, S.X., Simpson, M.F., Phongikaroon, S. (2006). Electrolytic reduction of spent nuclear oxide fuel as part of an integral process to separate and recover actinides from fission products. *Separ. Sci. Technol.*, 41, 1965–1983.
36. Hesketh, K. (2012). A new non-proliferation assessment tool. *Nucl. Eng. Int.*, 57, 36–40.
37. Hill, C. (2011). Development of highly selective compounds for solvent extraction processes: partitioning and transmutation of long-lived radionuclides from spent nuclear fuels. In K.L. Nash & G.J. Lumetta (Eds.), *Advanced Separation Techniques for Nuclear Fuel Reprocessing and Radioactive Waste Treatment* (pp. 311–362). Woodhead Publishing.
38. Ignatiev, V., Feynberg, O., Gnidoi, I., Merzlyakov, A., Surenkov, A., Uglov, V., Zagnitko, A., Subbotin, V., Sannikov, I., Toropov, A., Afonichkin, V., Bovet, A., Khokhlov, V., Shishkin, V., Kormilitsyn, M., Lizin, A., Osipenko, A. (2014). Molten salt actinide recycler and transforming system without and with Th–U support: fuel cycle flexibility and key material properties. *Ann. Nucl. Energy*, 64, 408–420.
39. Ikeda, K., Koyama, S.-i., Kurata, M., Morita, Y., Tsujimoto, K., Minato, K. (2014). Technology readiness assessment of partitioning and transmutation in Japan and issues toward closed fuel cycle. *Prog. Nucl. Energy*, 74, 242–263.
40. Ion, S. (2017). Challenges to deployment of twenty-first century nuclear reactor systems. *Proc. R. Soc. Math. Phys. Eng. Sci.*, 473.
41. Joly, P., Boo, E. (2015). *Roamap: Actinide Separation Processes 2015*. SACSESS, Paris.
42. Jonke, A.A. (1965). Reprocessing of nuclear reactor fuels by processes based on volatilization, fractional distillation, and selective adsorption. *Atom. Energy Rev.*, 3, 3–60.
43. Kani, Y., Sasahira, A., Hoshino, K., Kawamura, F. (2009). New reprocessing system for spent nuclear reactor fuel using fluoride volatility method. *J. Fluorine Chem.*, 130, 74–82.
44. Kato, T., Inoue, T., Iwai, T., Arai, Y. (2006). Separation behaviors of actinides from rare-earths in molten salt electrorefining using saturated liquid cadmium cathode. *J. Nucl. Mater.*, 357, 105–114.
45. Kim, Y.-I., Lee, H. (2015). 21 - development of closed nuclear fuel cycles in Korea. In R. Taylor (Ed.), *Reprocessing and Recycling of Spent Nuclear Fuel* (pp. 549–564). Woodhead Publishing.
46. Kobayashi, H., Amano, O., Kawamura, F., Aoi, M., Hoshino, K., Sasahira, A., Kani, Y. (2005). Fluorex reprocessing system for the thermal reactors cycle and future thermal/fast reactors (coexistence) cycle. *Prog. Nucl. Energy*, 47, 380–388.
47. König, A., Schreiner, A. (2001). Purification potential of melt crystallisation. *Powder Technol.*, 121, 88–92.
48. Koyama, T. (2011). 10 - nuclear engineering for pyrochemical treatment of spent nuclear fuels. In K.L. Nash & G.J. Lumetta (Eds.), *Advanced Separation Techniques for Nuclear Fuel Reprocessing and Radioactive Waste Treatment* (pp. 269–310). Woodhead Publishing.
49. Koyama, T., Iizuka, M. (2015). 18 - pyrochemical fuel cycle technologies for processing of spent nuclear fuels: developments in Japan. In R. Taylor (Ed.), *Reprocessing and Recycling of Spent Nuclear Fuel* (pp. 457–519). Woodhead Publishing.
50. Lacquement, J., Boussier, H., Laplace, A., Conocar, O., Grandjean, A. (2009). Potentialities of fluoride-based salts for specific nuclear reprocessing: overview of the R&D program at CEA. *J. Fluorine Chem.*, 130, 18–21.